



ALEXANDRIA

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Revista de Educação em Ciência e Tecnologia

MATERIAL SUPLEMENTAR

Referências dos estudos localizados (n = 401 materiais)

1) SL5

Parga-Lozano, D. L., & Carvalho, W. L. P. (2019). A pesquisa sobre ambientalização curricular. *Tecné, Episteme y Didaxis: TED*, (46), 39-56.

2) SL8

Cunha, S., Costa, O. B. S., Santana, L. L. B., & Lopes, W. A. (2015). Acetanilida: Síntese verde sem solvente. *Química Nova*, 38(6), 874-876.
<https://doi.org/10.5935/0100-4042.20150080>

3) SL9

Martinez, S. T., Silva, B. V., Pinto, A. C., Ferreira, V. F., & Silva, F. C (2012). Adição de anilinas à naftoquinona em água e em fase sólida. *Química Nova*, 35(4), 858-860. <https://doi.org/10.1590/S0100-40422012000400037>

4) SL11

Cunha, S., & Matos, J. S. (2017). Além da caipirinha: Cachaça como solvente para síntese orgânica e extração de pigmento. *Química Nova*, 40(10), 1253-1258.
<https://doi.org/10.21577/0100-4042.20170110>

5) SL16

Luz, L. T. S., Gomes, S. I. A. A., Sandri, M. C. M., Mello, F., & Bolzan, J. A. (2019). Avaliação e otimização das condições de obtenção do ácido acetilsalicílico para fins didáticos. *Educación Química*, 30(2), 54-69.
<https://doi.org/10.22201/fq.18708404e.2019.2.67393>

6) SL19

Cunha, S., Lustosa, D. M., Conceição, N. D., Fascio, M., & Magalhães, V. (2012).

Biomassa em aula prática de Química Orgânica Verde: Cravo-da-índia como fonte simultânea de óleo essencial e de furfural. *Química Nova*, 35(3), 638-641. <https://doi.org/10.1590/S0100-40422012000300035>

7) SL27

Cunha, S., & Santana, L. L. B. (2012). Condensação de Knoevenagel de aldeídos aromáticos com o ácido de Meldrum em água: Uma aula experimental de Química Orgânica Verde. *Química Nova*, 35(3), 642-647.

<https://doi.org/10.1590/S0100-40422012000300036>

8) SL28

Parga Lozano, D. L. (2015). Conhecimento Didático do Conteúdo sobre a Química Verde: O caso dos professores universitários de Química. *Tecné, Episteme y Didaxis: TED*, (38), 167-182.

<http://dx.doi.org/10.17227/01213814.38ted167.182>

9) SL29

Coelho, J. C., & Marques, C. A. (2007). Contribuições freireanas para a contextualização no ensino de Química. *Ensaio Pesquisa em Educação em Ciências*, 9(1), 59-75. <https://doi.org/10.1590/1983-21172007090105>

10) SL30

Pitanga, A. F. (2016). Crise da modernidade, Educação Ambiental, Educação para o Desenvolvimento Sustentável e Educação em Química Verde:(Re)pensando paradigmas. *Ensaio Pesquisa em Educação em Ciências*, 18(3), 141-159.

<https://doi.org/10.1590/1983-21172016180307>

11) SL35

Mello, F., Gomes, S. I. A. A., Giusti, E. D., Sandri, M. C. M., & Robaert, S. (2019).

Determinação do grau de saponificação de óleo residual: Uma experiência no ensino de Química sob as perspectivas CTSA e Química Verde. *Educación Química*, 30(1), 21-30. <https://doi.org/10.22201/fq.18708404e.2019.1.64110>

12) SL40

Machado, A. A. S. C. (2012). Dos primeiros aos segundos doze princípios da Química Verde. *Química Nova*, 35(6), 1250-1259.

<https://doi.org/10.1590/S0100-40422012000600034>

13) SL43

Vargas-Rodríguez, Y. M., Valdivia, A. O., Vargas, S. L., Escamilla, A. H., Ruvalcaba, R. M., & Rodríguez, G. I. V. (2016). El diagrama de flujo como semáforo de seguridad ecológica de los experimentos de laboratorio. *Educación Química*, 27(1), 30-36. <https://doi.org/10.1016/j.eq.2015.04.013>

14) SL45

González, M. B. R. (2018). Encadenamiento mediado por aprendizaje basado en proyectos ecoeficientes. *Sophia*, 14(2), 60-72. <https://doi.org/10.18634/sophiaj.14v.2i.788>

15) SL59

León-Cedeño, F. (2009). Implementation de algunas de las técnicas de la Química Verde (o Química Sustentable) endocencia. *Educación Química*, 20(4), 441-446. [https://doi.org/10.1016/S0187-893X\(18\)30048-X](https://doi.org/10.1016/S0187-893X(18)30048-X)

16) SL61

Merat, L. M. O. C., & Gil, R. A. S. S. (2003). Inserção do conceito de economia atômica no programa de uma disciplina de Química Orgânica Experimental. *Química Nova*, 26(5), 779-781. <https://doi.org/10.1590/S0100-40422003000500025>

17) SL64

Miranda, R., Noguez, O., Velasco, B., Arroyo, G., Penieres, G., Martínez, J. O., & Delgado, F. (2009). Irradiación infrarroja: Una alternativa para la activación de reacciones y su contribución a la Química Verde. *Educación Química*, 20(4), 421-425.

18) SL68

Ayuso, A. H. (2009). Los estudios de posgrado en Química Sostenible en España. *Educación Química*, 20(4), 405-411. [https://doi.org/10.1016/S0187-893X\(18\)30043-0](https://doi.org/10.1016/S0187-893X(18)30043-0)

19) SL78

Bruno, C. M. A., & Almeida, M. R. (2021). Óleos essenciais e vegetais: Matérias-primas para fabricação de bioprodutos nas aulas de Química Orgânica Experimental. *Química Nova*, 44(7), 899-907. <https://doi.org/10.21577/0100-4042.20170722>

20) SL82

Sandri, M. C. M., & Santin Filho, O. (2019). Os modelos de abordagem da Química Verde no ensino de Química. *Educación Química*, 30(4), 34-46.

<https://doi.org/10.22201/fq.18708404e.2019.4.68335>

21) SL83

Santos, A. P. B., Gonçalves, I. R. C., Pais, K. C., Martinez, S. T., Lachter, E. R., & Pinto, A. C. (2009). Oxidação do borneol à cânfora com água sanitária-um experimento simples, de baixo custo e limpo. *Química Nova*, 32(6), 1667-1669.

<https://doi.org/10.1590/S0100-40422009000600053>

22) SL84

Succaw, G. L., & Doxsee, K. M. (2009). Palladium-catalyzed synthesis of a benzofuran: A case study in the development of a Green Chemistry laboratory experiment. *Educación Química*, 20(4), 433-440.

[https://doi.org/10.1016/S0187-893X\(18\)30047-8](https://doi.org/10.1016/S0187-893X(18)30047-8)

23) SL93

Maximiano, F. A., Corio, P., Porto, P. A., & Fernandez, C. (2009). Química Ambiental e Química Verde no conjunto do conhecimento químico: Concepções de alunos de graduação em Química da Universidade de São Paulo. *Educación Química*, 20(4), 398-404.

[https://doi.org/10.1016/S0187-893X\(18\)30042-9](https://doi.org/10.1016/S0187-893X(18)30042-9)

24) SL96

Wallau, M., Bianchini, D., Ebersol, C. P., Santos Júnior, A. D., & Barboza, T. M. (2015). Química verdadeiramente verde - propriedades químicas do cloro e sua ilustração por experimentos em escala miniaturizada. *Química Nova*, 38(3), 436-445.

<https://doi.org/10.5935/0100-4042.20140314>

25) SL98

Fandiño, J. S. M., & Marín, L. A. E. (2021). Química Verde aplicada en los residuos de universidades. *Educación Química*, 32(2), 154-167.

<https://doi.org/10.22201/fq.18708404e.2021.2.76534>

26) SL102

Serrano, M. C. D. (2009). Química Verde: Un nuevo enfoque para el cuidado del medio ambiente. *Educación Química*, 20(4), 412-420.

[https://doi.org/10.1016/S0187-893X\(18\)30044-2](https://doi.org/10.1016/S0187-893X(18)30044-2)

27) SL106

Omori, A. T., Portas, V. B., & Oliveira, C. S. (2012). Redução enzimática do 4-(dimetilamino) benzaldeído com pedaços de cenoura (*Daucus carota*): Um experimento simples na compreensão da biocatálise. *Química Nova*, 35(2), 435-437. <https://doi.org/10.1590/S0100-40422012000200036>

28) SL107

Mendoza, M. S., Sánchez, A. V., Manrique, C. G., & Ávila-Zárraga, J. G. (2013). Reducción de nitrocompuestos utilizando el sistema Pd/H₂N-NH₂/Mw. *Educación Química*, 24(3), 347-350. [https://doi.org/10.1016/S0187-893X\(13\)72485-6](https://doi.org/10.1016/S0187-893X(13)72485-6)

29) SL108

Santos, R. V., Viana, G. M., Moreira, A. F. S., Nóbrega, V. S., Silva, V. A. S., Malta, L. F. B., Aguiar, L. C. S., & Senra, J. D. (2019). Revisiting the nucleophilicity concept in a comprehensive biomass valorization experiment: From papaya seeds to thiourea motifs. *Química Nova*, 42(8), 940-946. <https://doi.org/10.21577/0100-4042.20170395>

30) SL113

Cunha, S., Santos Filho, R. F., Riatto, V. B., & Dourado, G. A. A. (2013). Síntese e hidrólise de azalactonas de Erlenmeyer-Plöchl mediadas por radiação micro-ondas em aparelhos doméstico e dedicado: experimentos de Química Orgânica para a graduação. *Química Nova*, 36(1), 190-194. <https://doi.org/10.1590/S0100-40422013000100032>

31) SL114

Messina, L. C., & Omori, A. T. (2021). Síntese verde de 1, 3-diariltriazenos simétricos e assimétricos em vinagre. *Química Nova*, 44(3), 372-376. <https://doi.org/10.21577/0100-4042.20170655>

32) SL116

Ávila-Zárraga, J. G. (2009). Síntesis fotoquímica mediante luz solar. *Educación Química*, 20(4), 426-432. [https://doi.org/10.1016/S0187-893X\(18\)30046-6](https://doi.org/10.1016/S0187-893X(18)30046-6)

33) SL119

Vega Botto, A., & Torres, L. D. S. (2018). Sustitución de una práctica de Laboratorio con enfoque a "Química Verde" como herramienta para

la reducción de residuos peligrosos. *Educación Química*, 29(1), 110-120.

<https://doi.org/10.22201/fq.18708404e.2018.1.63809>

34) SL127

Mansilla, D. S., Muscia, G. C., & Ugliarolo, E. A. (2014). Una fundamentación para la incorporación de la Química Verde en los currículos de Química Orgánica.

Educación Química, 25(1), 56-59. [https://doi.org/10.1016/S0187-893X\(14\)70524-5](https://doi.org/10.1016/S0187-893X(14)70524-5)

35) SL131

Aragão, N. M., Veloso, M. C. C., & Andrade, J. B. (2009). Validação de métodos cromatográficos de análise: um experimento de fácil aplicação utilizando cromatografia líquida de alta eficiência (CLAE) e os princípios da "Química Verde" na determinação de metilxantinas em bebidas. *Química Nova*, 32(9), 2476-2481. <https://doi.org/10.1590/S0100-40422009000900043>

36) E2

Hopson, R., Lee, P. Y. D., & Hess, K. M. (2018). 1-Dimensional selective nuclear overhauser effect NMR spectroscopy to characterize products from a two-step Green Chemistry synthesis. *Journal of Chemical Education*, 95(4), 641-647. <https://doi.org/10.1021/acs.jchemed.7b00494>

37) E3

Phonchaiya, S., Panijpan, B., Rajviroongit, S., Wright, T., & Blanchfield, J. T. (2009). A facile solvent-free cannizzaro reaction. *Journal of Chemical Education*, 86(1), 85-86. <https://doi.org/10.1021/ed086p85>

38) E4

Bennett, G. D. (2006). A green enantioselective aldol condensation for the undergraduate organic laboratory. *Journal of Chemical Education*, 83(12), 1871-1872. <https://doi.org/10.1021/ed083p1871>

39) E5

Hooper, M. M., & Boef, B. (2009). A green multicomponent reaction for the Organic Chemistry laboratory: The aqueous Passerini reaction. *Journal of Chemical Education*, 86(9), 1077-1079. <https://doi.org/10.1021/ed086p1077>

40) E7

Bennett, G. D. (2005). A green polymerization of aspartic acid for the undergraduate organic laboratory. *Journal of Chemical Education*, 82(9), 1380-1381.
<https://doi.org/10.1021/ed082p1380>

41) E8

Jones-Wilson, T. M., & Burtch, E. A. (2005). A green starting material for electrophilic aromatic substitution for the undergraduate organic laboratory. *Journal of Chemical Education*, 82(4), 616-617. <https://doi.org/10.1021/ed082p616>

42) E11

Lazarski, K. E., Rich, A. A., & Mascarenhas, C. M. (2008). A one-pot, asymmetric Robinson annulation in the Organic Chemistry majors laboratory. *Journal of Chemical Education*, 85(11), 1531-1534. <https://doi.org/10.1021/ed085p1531>

43) E12

Bopegedera, A. M. R. P. (2018). A second look at the kinetics of the Iron–Oxygen reaction: Determination of the total order using a greener approach. *Journal of Chemical Education*, 95(10), 1897-1899.
<https://doi.org/10.1021/acs.jchemed.8b00012>

44) E15

Barcena, H., & Chen, P. (2016). An anesthetic drug demonstration and an introductory antioxidant activity experiment with “Eugene, the sleepy fish”. *Journal of Chemical Education*, 93(1), 202-205.
<https://doi.org/10.1021/ed5008469>

45) E19

Owens, J. E., Zimmerman, L. B., Gardner, M. A., & Lowe, L. E. (2016). Analysis of whiskey by dispersive liquid–liquid microextraction coupled with gas chromatography/mass spectrometry: An upper division analytical chemistry experiment guided by green chemistry. *Journal of Chemical Education*, 93(1), 186-192. <https://doi.org/10.1021/acs.jchemed.5b00342>

46) E20

O’Neil, N. J., Scott, S., Relph, R., & Ponnusamy, E. (2021). Approaches to incorporating Green Chemistry and safety into laboratory culture. *Journal of Chemical Education*, 98(1), 84-91.
<https://doi.org/10.1021/acs.jchemed.0c00134>

47) E22

Ribeiro, M. G. T. C., Yunes, S. F., & Machado, A. A. S. C. (2014). Assessing the greenness of chemical reactions in the laboratory using updated holistic graphic metrics based on the globally harmonized system of classification and labeling of chemicals. *Journal of Chemical Education*, 91(11), 1901-1908. <https://doi.org/10.1021/ed400421b>

48) E25

Santos, E. S., Garcia, I. C. G., & Gomez, E. F. L. (2004). Caring for the environment while teaching Organic Chemistry. *Journal of Chemical Education*, 81(2), 232-238. <https://doi.org/10.1021/ed081p232>

49) E26

Cerrillo, J. L., Lopez-Hernandez, I., & Palomares, A. E. (2021). Catalytic removal of bromates from water: A hands-on laboratory experiment to solve a water pollution problem through catalysis. *Journal of Chemical Education*, 98(5), 1726-1731. <https://doi.org/10.1021/acs.jchemed.0c01297>

50) E28

Cunningham, A. D., Ham, E. Y., & Vosburg, D. A. (2011). Chemoselective reactions of citral: Green syntheses of natural perfumes for the undergraduate organic laboratory. *Journal of Chemical Education*, 88(3), 322-324. <https://doi.org/10.1021/ed100539m>

51) E33

Fennie, M. W., & Roth, J. M. (2016). Comparing amide-forming reactions using green chemistry metrics in an undergraduate organic laboratory. *Journal of Chemical Education*, 93(10), 1788-1793. <https://doi.org/10.1021/acs.jchemed.6b00090>

52) E34

Cacciatore, K. L., Amado, J., Evans, J. J., & Sevian, H. (2008). Connecting solubility, equilibrium, and periodicity in a green, inquiry experiment for the general chemistry laboratory. *Journal of Chemical Education*, 85(2), 251-253. <https://doi.org/10.1021/ed085p251>

53) E41

Armenta, S., & Guardia, M. (2011). Determination of mercury in milk by cold vapor atomic fluorescence: A Green Analytical Chemistry laboratory experiment.

Journal of Chemical Education, 88(4), 488-491.

<https://doi.org/10.1021/ed100578g>

54) E43

Akers, S. M., Conkle, J. L., Thomas, S. N., & Rider, K. B. (2006). Determination of the heat of combustion of biodiesel using bomb calorimetry. A multidisciplinary undergraduate chemistry experiment. *Journal of Chemical Education*, 83(2), 260-262. <https://doi.org/10.1021/ed083p260>

55) E48

Aubrecht, K. B., Padwa, L., Shen, X., & Bazargan, G. (2015). Development and implementation of a series of laboratory field trips for advanced high school students to connect chemistry to sustainability. *Journal of Chemical Education*, 92(4), 631-637. <https://doi.org/10.1021/ed500630f>

56) E54

Sjöström, J. (2013). Eco-driven chemical research in the boundary between academia and industry: PhD students' views on science and society. *Science & Education*, 22, 2427-2441. <https://doi.org/10.1007/s11191-012-9490-4>

57) E57

Schultz, M. (2013). Embedding environmental sustainability in the undergraduate chemistry curriculum: A case study. *Journal of Learning Design*, 6(1), 20-33.

58) E58

Ravía, S., Gaménara, D., Schapiro, V., Bellomo, A., Adum, J., Seoane, G., & Gonzalez, D. (2006). Enantioselective reduction by crude plant parts: Reduction of benzofuran-2-yl methyl ketone with carrot (*Daucus carota*) bits. *Journal of Chemical Education*, 83(7), 1049-1051. <https://doi.org/10.1021/ed083p1049>

59) E60

Mohan, R. S., & Mejia, M. P. (2020). Environmentally friendly organic chemistry laboratory experiments for the undergraduate curriculum: a literature survey and assessment. *Journal of Chemical Education*, 97(4), 943-959. <https://doi.org/10.1021/acs.jchemed.9b00753>

60) E65

Cósio, M. N., Cardenal, A. D., Maity, A., Hyun, S. M., Akwaowo, V. E., Hoffman, C. W., Poderes, T. M., & Powers, D. C. (2020). Exploring Green Chemistry with aerobic hypervalent iodine catalysis. *Journal of Chemical Education*, 97(10), 3816-3821. <https://doi.org/10.1021/acs.jchemed.0c00410>

61) E68

Hartwell, S. K. (2012). Exploring the potential for using inexpensive natural reagents extracted from plants to teach chemical analysis. *Chemistry Education Research and Practice*, 13(2), 135-146. <https://doi.org/10.1039/C1RP90070F>

62) E74

Divya, D., & Raj, K. G. (2019). From scrap to functional materials: Exploring Green and Sustainable Chemistry approach in the undergraduate laboratory. *Journal of Chemical Education*, 96(3), 535-539. <https://doi.org/10.1021/acs.jchemed.8b00484>

63) E76

McAllister, G. D., & Parsons, A. F. (2019). Going green in process chemistry: optimizing an asymmetric oxidation reaction to synthesize the antiulcer drug esomeprazole. *Journal of Chemical Education*, 96(11), 2617-2621. <https://doi.org/10.1021/acs.jchemed.9b00350>

64) E77

Haack, J. A., Hutchison, J. E., Kirchhoff, M. M., & Levy, I. J. (2005). Going green: Lecture assignments and lab experiences for the college curriculum. *Journal of Chemical Education*, 82(7), 974-976. <https://doi.org/10.1021/ed082p974>

65) E78

Leslie, J. M., & Tzeel, B. A. (2016). Gold (III)-catalyzed hydration of phenylacetylene. *Journal of Chemical Education*, 93(6), 1100-1102. <https://doi.org/10.1021/acs.jchemed.5b00628>

66) E81

Hjeresen, D. L., Schutt, D. L., & Boese, J. M. (2000). Green Chemistry and education. *Journal of Chemical Education*, 77(12), 1543-1547. <https://doi.org/10.1021/ed077p1543>

67) E91

Sutheimer, S., Caster, J. M., & Smith, S. H. (2015). Green soap: an extraction and saponification of avocado oil. *Journal of Chemical Education*, 92(10), 1763-1765. <https://doi.org/10.1021/acs.jchemed.5b00188>

68) E95

McKenzie, L. C., Huffman, L. M., Hutchison, J. E., Rogers, C. E., Goodwin, T. E., & Spessard, G. O. (2009). Greener solutions for the Organic Chemistry teaching lab: Exploring the advantages of alternative reaction media. *Journal of Chemical Education*, 86(4), 488-493. <https://doi.org/10.1021/ed086p488>

69) E96

Sidhwani, I. T., & Chowdhury, S. (2008). Greener alternative to qualitative analysis for cations without H₂S and other sulfur-containing compounds. *Journal of Chemical Education*, 85(8), 1099-1101. <https://doi.org/10.1021/ed085p1099>

70) E98

Karpudewan, M., Ismail, Z. H., & Mohamed, N. (2011). Greening a chemistry teaching methods course at the school of educational studies, Universiti Sains Malaysia. *Journal of Education for Sustainable Development*, 5(2), 197-214. <https://doi.org/10.1177/097340821100500210>

71) E100

Gron, L. U., Bradley, S. B., McKenzie, J. R., Shinn, S. E., & Teague, M. W. (2013). How to recognize success and failure: Practical assessment of an evolving, first-semester laboratory program using simple, outcome-based tools. *Journal of Chemical Education*, 90(6), 694-699. <https://doi.org/10.1021/ed200523w>

72) E103

Cann, M. C., & Dickneider, T. A. (2004). Infusing the chemistry curriculum with green chemistry using real-world examples, web modules, and atom economy in Organic Chemistry courses. *Journal of Chemical Education*, 81(7), 977-980. <https://doi.org/10.1021/ed081p977>

73) E105

Zuin, V. G., Segatto, M. L., Zandonai, D. P., Grosseli, G. M., Stahl, A., Zanotti, K., & Andrade, R. S. (2019). Integrating Green and Sustainable Chemistry into undergraduate teaching laboratories: Closing and assessing the loop on the basis of a citrus biorefinery approach for the biocircular economy in Brazil.

Journal of Chemical Education, 96(12), 2975-2983.

<https://doi.org/10.1021/acs.jchemed.9b00286>

74) E109

Zhu, J., Zhang, M., & Liu, Q. (2008). Interdisciplinary chemistry experiment: An environmentally benign extraction of lycopene. *Journal of Chemical Education*, 85(2), 256-257. <https://doi.org/10.1021/ed085p256>

75) E112

Kiryak, Z., Candas, B., & Özmen, H. (2021). Investigating preservice science teachers' cognitive structures on environmental issues. *Journal of Science Learning*, 4(3), 244-256. <https://doi.org/10.17509/jsl.v4i3.30366>

76) E113

Stark, A., Ott, D., Kralisch, D., Kreisel, G., & Ondruschka, B. (2010). Ionic liquids and Green Chemistry: A lab experiment. *Journal of Chemical Education*, 87(2), 196-201. <https://doi.org/10.1021/ed8000396>

77) E119

Tundo, P., Rosamilia, A. E., & Arico, F. (2010). Methylation of 2-naphthol using dimethyl carbonate under continuous-flow gas-phase conditions. *Journal of Chemical Education*, 87(11), 1233-1235. <https://doi.org/10.1021/ed100448y>

78) E120

Das, K. R., & Antony, M. J. (2022). Microscale redox titrations using poly-N-phenyl anthranilic acid fluorescent turn-off indicator for undergraduate Analytical Chemistry lab. *Journal of Chemical Education*, 99(2), 892-901. <https://doi.org/10.1021/acs.jchemed.0c01354>

79) E121

Uffelman, E. S., Doherty, J. R., Schultze, C., Burke, A. L., Bonnema, K. R., Watson, T. T., & Lee, D. W. (2004). Microscale syntheses, reactions, and ¹H NMR spectroscopic investigations of square planar macrocyclic Tetraamido-N Cu (III) complexes relevant to Green Chemistry. *Journal of Chemical Education*, 81(2), 182-185. <https://doi.org/10.1021/ed081p182>

80) E123

Reilly, M. K., King, R. P., Wagner, A. J., & King, S. M. (2014). Microwave-assisted esterification: A discovery-based microscale laboratory experiment. *Journal of Chemical Education*, 91(10), 1706-1709. <https://doi.org/10.1021/ed400721p>

81) E124

Dintzner, M. R., Wucka, P. R., & Lyons, T. W. (2006). Microwave-assisted synthesis of a natural insecticide on basic montmorillonite K10 Clay. Green Chemistry in the undergraduate organic laboratory. *Journal of Chemical Education*, 83(2), 270-272. <https://doi.org/10.1021/ed083p270>

82) E127

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